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stoichiometry (which you studied in Chapter 3) deals with the mass rela-tionships of elements in compoundsReaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction Reaction stoichiometry is the subject of this chapter and it is based on

Chapter 9 - Stoichiometry

Chapter 9 - Stoichiometry 9-1 Introduction to Stoichiometry Composition Stoichiometry - deals with mass relationships of elements in compounds Reaction Stoichiometry - Involves mass relationships between reactants and products in a chemical reaction I Reaction Stoichiometry Problems A Four problem Types, One Common Solution

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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g Calculate the percentage yield 2 60 mol of N 2 are mixed with 120 mol of H 2 according to the following equation: N 2(g) 3H 2(g

Chapter 03 - Stoichiometry

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